organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Peter G. Jones,^a* Anthony J. Kirby^b and Melanie Pilkington^b

^aInstitut für Anorganische und Analytische Chemie, Technische Universität Braunschweig, Postfach 3329, 38023 Braunschweig, Germany, and ^bUniversity Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW, England

Correspondence e-mail: jones@xray36.anchem.nat.tu-bs.de

Key indicators

Single-crystal X-ray study T = 143 K Mean σ (C–C) = 0.004 Å Disorder in main residue R factor = 0.048 wR factor = 0.119 Data-to-parameter ratio = 13.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The title compound (2,2,2-tris(cyclohexyloxy)phenanthro-[9,10-d][1,3,2]dioxaphosphole), $C_{32}H_{41}O_5P$, displays trigonalbipyramidal geometry at the central P atom. The P–O bond lengths (Å) are: in the five-membered ring 1.652 (2) Å equatorial and 1.744 (2) Å axial, and to the cyclohexyloxy groups 1.581 (2) and 1.589 (2) Å equatorial, and 1.624 (2) Å axial.

1,3,2-dioxaphospholene

2,2,2-Tris(cyclohexyloxy)-4,5-(2',2"-biphenylo)-

Comment

We are interested in the correlation between structure and reactivity (see *e.g.* Amos *et al.*, 1992). The structure of the title compound, (I), the adduct of tricyclohexyl phosphite and phenanthraquinone, was determined as part of a combined crystallographic and theoretical investigation of pKa values of pentaoxyphosphoranes (Davies *et al.*, 2002).



The structure is shown in Fig. 1; a minor (11%) disorder component of the ring C11-16 is not shown. Of the three cyclohexyloxy groups, one is axial (O5, C31 etc.) and the other two are equatorial. This is reflected in the normal bond-length pattern for trigonal-bipyramidal geometry (axial bonds longer than equatorial), as was also observed in pentakis(cyclohexyloxy)phosphorane (Davies et al., 2001; average values in four independent molecules 1.662 Å for axial and 1.606 Å for equatorial P-O bonds). The triisopropyloxy analogue of the title compound was subjected to structure analysis in 1967. There were two modifications, of which the more precisely determined modification (Spratley et al., 1967) displayed a similar P-O bond-length pattern to the title compound: ring apical 1.751 (7) Å, equatorial 1.633 (7) Å; i Pr apical 1.638 (6) Å, and equatorial 1.588 (7) and 1.574 (7) Å.

Received 1 February 2002 Accepted 5 February 2002 Online 22 February 2002

 \odot 2002 International Union of Crystallography Printed in Great Britain – all rights reserved

Peter G. Jones et al. • C₃₂H₄₁O₅P

0268



Figure 1

The molecule of the title compound in the crystal. The minor disorder component of the ring C11–C16 has been omitted. Ellipsoids represent 50% probability levels.

Experimental

Compound (I) was prepared by adding triisopropyl phosphite to one equivalent of phenanthraquinone at room temperature, according to the method of Hamilton *et al.* (1967).

Crystal data

 $\begin{array}{l} C_{32}H_{41}O_5P\\ M_r = 536.62\\ \text{Monoclinic, } P_{2_1}/n\\ a = 9.665 \ (2) \ \text{\AA}\\ b = 15.394 \ (3) \ \text{\AA}\\ c = 19.200 \ (4) \ \text{\AA}\\ \beta = 93.92 \ (2)^{\circ}\\ V = 2850.1 \ (10) \ \text{\AA}^3\\ Z = 4 \end{array}$

Data collection

Stoe Stadi-4 diffractometer ω/θ scans Absorption correction: none 5588 measured reflections 5037 independent reflections 3708 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.024$ $D_x = 1.251 \text{ Mg m}^{-3}$ Mo K α radiation Cell parameters from 50 reflections $\theta = 10-11.5^{\circ}$ $\mu = 0.14 \text{ mm}^{-1}$ T = 143 (2) K Prism, colourless $0.65 \times 0.30 \times 0.30 \text{ mm}$

 $\begin{aligned} \theta_{\max} &= 25.0^{\circ} \\ h &= -11 \rightarrow 0 \\ k &= -18 \rightarrow 1 \\ l &= -22 \rightarrow 22 \\ 3 \text{ standard reflections} \\ \text{frequency: 60 min} \\ \text{intensity decay: none} \end{aligned}$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0447P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.048$	+ 1.4014P]
$vR(F^2) = 0.119$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.04	$(\Delta/\sigma)_{\rm max} < 0.001$
037 reflections	$\Delta \rho_{\rm max} = 0.21 \text{ e } \text{\AA}^{-3}$
68 parameters	$\Delta \rho_{\rm min} = -0.31 \text{ e } \text{\AA}^{-3}$
I-atom parameters constrained	

Table 1

\$

5

ł

Selected geometric parameters (Å, °).

P-O3	1.581 (2)	P-O2	1.652 (2)
P-O4	1.589 (2)	P-O1	1.744 (2)
P-O5	1.624 (2)		
O3-P-O4	117.35 (9)	O5-P-O2	88.52 (8)
O3-P-O5	92.73 (9)	O3-P-O1	94.57 (8)
O4-P-O5	91.98 (8)	O4-P-O1	84.38 (8)
O3-P-O2	115.31 (9)	O5-P-O1	172.69 (9)
O4-P-O2	127.24 (9)	O2-P-O1	88.66 (8)

H atoms were positioned geometrically and refined using a riding model. Cyclohexyl ring C11–16 is disordered over two positions with a relative occupancy of ca 9:1. The minor component was refined isotropically with an appropriate system of restraints.

Data collection: *DIF*4 (Stoe & Cie, 1992); cell refinement: *DIF*4; data reduction: *REDU*4 (Stoe & Cie, 1992); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1990); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *XP* (Siemens, 1994); software used to prepare material for publication: *SHELXL*97.

Financial support from the Fonds der Chemischen Industrie is gratefully acknowledged. We thank Mr A. Weinkauf for technical assistance.

References

- Amos, R. D., Handy, N. C., Jones, P. G., Kirby, A. J., Parker, J. K., Percy, J. M. & Su, M. D. (1992). J. Chem. Soc. Perkin Trans 2, pp. 549–558.
- Davies, J. E., Kirby, A. J. & Roussev, C. D. (2001). Acta Cryst. E57, 0994-0995.
- Hamilton, W. C., LaPlaca, S. J., Ramirez, F. & Smith, C. P. (1967). J. Am. Chem. Soc. 89, 2268–2272.
- Davies, J. E., Doltsinis, N. L., Kirby, A. J., Roussev, C. D. & Sprik, M. (2002). J. Am. Chem. Soc. Submitted.
- Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.

Siemens (1994). XP. Version 5.03. Siemens Analytical X-ray Instruments, Madison, USA.

- Spratley, R. D., Hamilton, W. C. & Ladell, J. (1967). J. Am. Chem. Soc. 89, 2272–2278.
- Stoe & Cie (1992). DIF4 and REDU4. Stoe & Cie, Darmstadt, Germany.